

## Global N and P cycles

### I. Intro

- A. look at aspects of these cycles
- B. dominant processes and how man's activities are affecting them

### II. Nitrogen cycle

**Fig. 12.1**

- A. strongly coupled to the carbon cycle
  - 1. some key differences
- B. a larger variety of nitrogen species
  - 1.  $N_2$  - large reservoir in the atmosphere but not easily available to biota
    - a. other gaseous form - NO and  $N_2O$
  - 2. biologically available N - ammonium, nitrite, nitrate
  - 3. nitrogen oxides such as  $N_2O$ 
    - a.  $N_2O$  not generally thought of as a reactive form of N but it is a greenhouse gas
    - b. produced by both nitrification and denitrification
    - c. other nitrogen oxides
  - 4. a large number of biogeochemical transformations possible

### III. Global nitrogen cycle

**Reeburg figure**

- A. atmospheric  $N_2$  is the largest pool
  - 1.  $\sim 10^9$  Tg N in atmos. vs.  $\sim 10^5$  g in terr systems (terr. biomass and soil OM)
  - 2. oceans contain  $\sim 10^6$  g of "bioactive" N (not  $N_2$ )
- B. contrasts other elements that have their largest pool in sed rocks (ie, C)
  - 1. instead of burying it away for long periods of time, N cycling "stores" most N in the atmos. in a form that is simply much less biologically available
  - 2. atmospheric residence time is over 20 million years
- C. Human activity has severely affected global N cycle
  - 1. first look at controls on pre-industrial N cycle
  - 2. then look at how human activity is affecting the cycle

### IV. Long-term N cycle dominated by a balance between biological N-fixation and denitrification

- A. rapid internal cycling both in the oceans and on land **Fig. 12.1**
- B. these may not always be balanced – **show my cartoon**
- C. some studies suggest that the pre-industrial N budget was not in steady-state
  - 1. pre-industrial ocean may have been losing fixed nitrogen
  - 2. opposite may have been the case during glacial times
- D. evidence that combined N imbalance may change on glacial/interglacial time scales
  - 1. decrease in denitrification rates during glacial times
    - a. possibly related to drop in sea level and decrease in the area of continental shelves (important sites of denitrification)
    - b. could help explain the lower atmos  $CO_2$  during glacial times **show my systems diagram**
  - 2. could also be related to an increase in N fixation **Fig. 14-13**

### V. Man's impact on the nitrogen cycle

- A. human activities have severely affected the global N cycle

B. add fixed N in several ways

show N cascade figure

1. energy production - fossil fuel burning
  - a. converts  $N_2$  and org N in fossil fuels to NO
  - b. NO oxidized to  $NO_2$  which then rains out as nitrate
    - (i) leads to acid rain
    - (ii) has a rel. short lifetime in atmosphere
    - (iii) see atmos. Fixed N turnover times in **Fig. 3**
2. fertilizer production - Haber process
  - a. converts  $N_2$  to ammonium
3. crop cultivation
  - a. human-induced increase in biological nitrogen fixation

VI. Anthropogenic cycles of reactive N

**Fig. 3 Reeburg**

- A. on continents, human activities are producing reactive N at comparable rates to natural processes
  1. ~140 vs. ~190 Tg N/yr
- B. release of reactive N from continents to coastal oceans has doubled **Fig. 2**
  1. comparison of pre-ind. world to present day
    - a. present day = pre-ind. + anthropogenic fluxes
- C. net atm transfer of N from continents to oceans has gone from ~0 to 18 Tg N/yr
  1. absolute emission of reactive N to atmos have increased
    - a. 29 Tg/yr (pre-ind) to 107 Tg/yr (currently)
- D. allows more reactive N to be re-distributed to points far from the pt. of fixation **Fig. 12.3**

VII. human activity on land fixes ~ 140 Tg N/yr

**Fig. 2 with animation**

- A. where does it go
- B. 41 Tg/yr transported to coastal ocean
- C. 18 Tg/yr are deposited on oceans from atmos.
  1. primarily from atmos deposition
- D. 3.4 Tg/yr are emitted to atmos as  $N_2O$ 
  1. leads to an increase in atmos.  $N_2O$  increasing
  2.  $N_2O$  less reactive in atmos. longer turnover time
    - a. another greenhouse gas
- E. ~80 Tg/yr not accounted for
  1. retained on land in groundwater, soils or vegetation
  2. may be denitrified on land
    - a. estimate of this is 50 -110 Tg/yr
    - b. based on  $N_2O$  prdn. rate and a  $N_2:N_2O$  denitrif. ratio of 14-32

VIII. Fate Anthro reactive N added to oceans

- A. 41 Tg/yr transported to coastal ocean by rivers
- B. 18 Tg/yr are deposited on oceans from atmos.
- C. What is its fate **Fig. 6**
  1. buried in seds (some) or denitrified (most)
  2. in a local sense this plays a role in cultural eutrophication of coastal regions
  3. little of this apparently escapes coastal ocean

- a. no enhanced  $C_{org}$  burial in open ocean, implies that N burial has not increased
- b. DN on shelves may be greater than river input of reactive N
- D. on land and in the ocean the ultimate fates of anthro reactive nitrogen is denitrification
  - 1. the problem is what happens along the way

#### **IX. Impacts of increasing anthropogenic N fixation**

**go through Nr slides**

- 1. once an atom of Nr enters the system it can cause multiple effects
  - a. in the atmosphere, in terr. ecosystem and in marine and freshwater environments
- 2. effects become magnified over time
- B. Cascade of reactive N
  - 1. Reactive nitrogen (Nr) is accumulating in the atmosphere.
  - 2. Nr is accumulating in terrestrial systems
  - 3. Once Nr enters water (streams, rivers, estuaries) most will eventually denitrify
    - a. generally far from point of entry.
  - 4. Accumulation occurs on short- and long-time scales
    - a. different reservoirs have different residence times
    - b. leads to accumulation

**go thru N cascade figures**

#### **X. Forecast for the future**

- A. anthro N fixation driven by energy and food production
  - 1. controlled by human pop. and standards of living
  - 2. both will prob. incr. with time
- B. the implications of this increase will depend on how and where this N accumulates
- C. even if most is eventually denitrified there are still regional and global problems

**EHP Fig.**

#### **XI. Phosphorus cycle**

- A. linked to N & C cycles
  - 1. some distinct diffs in P cycle
  - 2. P has no significant atmos component
  - 3. phosphate and org. P are the predom. species
- B. much more of a “geological” cycle
  - 1. no significant atmospheric fluxes from continent to oceans
    - a. atmospheric inventory component minor
  - 2. on long time scales, P is generally considered to be the major nutrient limiting primary prod.
- C. internal recycling very important – particularly in the oceans
- D. Increased P remobilization by human activity
  - 1. mineable P in the budget
  - 2. see it, for example, in incr. P loading by rivers
  - 3. again its effect will be more regional rather than global (like N)
    - a. no major atmos. P cycle
    - b. P cycling relies strongly on hydrologic transport
  - 4. distribution from fertilizer run-off and sewage treatment is likely confined to freshwater and nearshore coastal ecosystems

**look at figures**

**Fig. 10.16 Mackenzie**